



Canadian Intellectual
Property Office

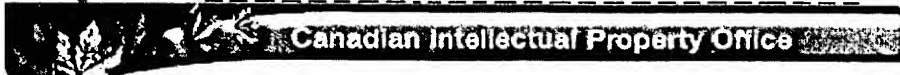
Office de la propriété
intellectuelle du Canada

Canada

Français	Contact Us	Help	Search	Canada Site
Strategis	Site Map	What's New	About Us	Registration



Strategis Index: [A](#) [B](#) [C](#) [D](#) [E](#) [F](#) [G](#) [H](#) [I](#) [J](#) [K](#) [L](#) [M](#) [N](#) [O](#) [P](#) [Q](#) [R](#) [S](#) [T](#) [U](#) [V](#) [W](#) [X](#) [Y](#) [Z](#)



Canadian Patents Database

(12) Patent:

(11) CA 2103595

(54) POLYMERISATION REGULATION

(54) REGULATION DE POLYMERISATION

[View or Download Images](#)

[View Administrative Status](#)

ABSTRACT:

A process for the production of low molecular weight polymers by free radical polymerisation of one or more monomers in the presence of a chain transfer agent, characterized in that the chain transfer agent comprises one or more compounds of general formula (1), wherein R1 is a hydrogen atom, an alkyl group, or a group capable of activating the vinylic carbon towards free radical addition; R2 represents an optionally substituted alkyl, optionally substituted alkenyl, optionally substituted alkynyl, or optionally substituted saturated, unsaturated or aromatic carbocyclic or heterocyclic ring; and Z represents an oxygen, sulphur, phosphorus, or nitrogen atom bound to another atom or group of atoms in order to satisfy its valency.

CLAIMS: [Show all claims](#)

*** Note: Data on abstracts and claims is shown in the official language in which it was submitted.

(72) Inventors (Country): **MEIJS, GORDON FRANCIS** (Australia)
RIZZARDO, EZIO (Australia)

(73) Owners (Country): **COMMONWEALTH SCIENTIFIC AND INDUSTRIAL
RESEARCH ORGANISATION** (Australia)

(71) Applicants (Country): **COMMONWEALTH SCIENTIFIC AND INDUSTRIAL
RESEARCH ORGANISATION** (Australia)

(74) Agent: **DIMOCK STRATTON CLARIZIO LLP**

(45) Issued: **Aug. 20, 2002**

(86) PCT Filing Date: **Jan. 31, 1992**

(87) PCT Publication Date: **Aug. 20, 1992**

Examination requested: **Jan. 26, 1999**

(51) International Class (IPC): **C08F 2/38**

Patent Cooperation Treaty (PCT): **Yes**

(85) <u>National Entry</u> :	Aug. 6 , 1993
(86) <u>PCT Filing number</u> :	PCT/AU1992/000029
(87) <u>International publication number</u> :	WO1992/013903

(30) Application priority data:

Application No.	Country	Date
PK4465/91	Australia	Feb. 6 , 1991

Availability of licence: **N/A**

Language of filing: **English**

View or Download Images :

- ☐ Cover Page Image
- ☐ Abstract Image
- ☐ Claims Image
- ☐ Disclosures Image
- ☐ Drawings Image
- ☐ Representative Drawing Image

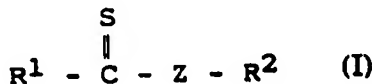
[View the Image](#)

[Download in Adobe PDF](#)

WORLD INTELLECTUAL PROPERTY ORGANIZATION
International Bureau

(51) International Patent Classification ⁵ : C08F 002/38, C07C 327/26	(11) International Publication Number: WO 92/13903
2103595 ^{A1}	(43) International Publication Date: 20 August 1992 (20.08.92)
(21) International Application Number: PCT/AU92/00029	(74) Agents: CORBETT, Terence, Guy et al.; Davies Collison
(22) International Filing Date: 31 January 1992 (31.01.92)	Cave, 1 Little Collins Street, Melbourne, VIC 3000 (AU).
(30) Priority data: PK 4465 6 February 1991 (06.02.91) AU	(81) Designated States: AT (European patent), AU, BE (Euro-
(71) Applicant (for all designated States except US): COMMON-	pean patent), CA, CH (European patent), DE (Euro-
WEALTH SCIENTIFIC AND INDUSTRIAL RE-	pean patent), DK (European patent), ES (European pa-
SEARCH ORGANISATION [AU/AU]; Limestone	tent), FR (European patent), GB (European patent), GR
Avenue, Campbell, ACT 2601 (AU).	(European patent), IT (European patent), JP, LU (Euro-
(72) Inventors; and	pean patent), MC (European patent), NL (European pa-
(75) Inventors/Applicants (for US only): MEIJS, Gordon, Fran-	tent), SE (European patent), US.
cis [AU/AU]; 3 Henty Street, Murrumbidgee, VIC 3163	Published
(AU). RIZZARDO, Ezio [AU/AU]; 26 Alex Avenue,	With international search report.
Wheelers Hill, VIC 3150 (AU).	

(54) Title: POLYMERISATION REGULATION

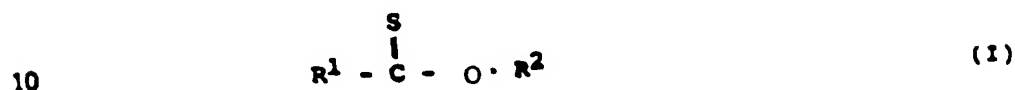


(57) Abstract

A process for the production of low molecular weight polymers by free radical polymerisation of one or more monomers in the presence of a chain transfer agent, characterized in that the chain transfer agent comprises one or more compounds of general formula (I), wherein R¹ is a hydrogen atom, an alkyl group, r a group capable of activating the vinylic carbon towards free radical addition; R² represents an optionally substituted alkyl, optionally substituted alkenyl, optionally substituted alkynyl, or optionally substituted saturated, unsaturated or aromatic carbocyclic or heterocyclic ring; and Z represents an oxygen, sulphur, phosphorus, r nitrogen atom bound to another atom or group of atoms in order to satisfy its valency.

CLAIMS

1. A process for the production of low molecular weight polymers by free radical polymerisation of one or more monomers in the presence of a chain transfer agent, characterised in that the chain transfer agent comprises one or more compounds of the general formula I



wherein

R^1 is a hydrogen atom, an alkyl group, or a group capable of activating the vinylic carbon towards free radical addition; and

15

R^2 represents an optionally substituted alkyl, optionally substituted alkenyl, optionally substituted alkynyl, or optionally substituted saturated, unsaturated or aromatic carbocyclic or heterocyclic ring; and

2. A process as claimed in Claim 1, characterised in that R^1 is a substituted phenyl or other optionally substituted aromatic group, or an alkoxycarbonyl or aryloxy carbonyl, carboxy, acyloxy, carbamoyl, or cyano group.

3. A process as claimed in Claim 1 or Claim 2, characterised in that R^1 and/or R^2 in formula I have one or more reactive substituent groups which do not take part in the actual lowering of the molecular weight but are installed at the ends of the polymer chains and may be capable of subsequent chemical reaction, whereby the low molecular weight polymer product containing the reactive group or groups is thereby able to undergo further chemical transformation.

4. A process as claimed in Claim 3, characterised in that the reactive

SUBSTITUTE SHEET

2103595

- 20 -

substituent group is an hydroxy, amin, halogen, allyl, cyano, epoxy, or carboxylic acid group or a derivative of a carboxylic acid group.

5. A process as claimed in any one of the preceding Claims, characterised in
5 that the compound of general formula I is added to present in an amount of from 0.01 to 30 mole percent based on total monomer.

6. A process as claimed in any one of the preceding Claims, characterised in
that the monomer is selected from the groups consisting of acrylic esters,
10 methacrylic esters, vinyl esters, vinyl aromatics, unsaturated or poly unsaturated hydrocarbons, or mixtures of any two or more of such monomers.

7. A process as claimed in any one of the preceding Claims, characterised in
that the resulting polymer is hydrolysed to give a terminal thiol group.
15

SUBSTITUTE SHEET

POLYMERISATION REGULATION

- The invention relates to processes for radical-initiated polymerisation of unsaturated species and for the regulation of molecular weight and end-group functionality of the polymeric products produced from such processes. Polymers of limited molecular weights, or oligomers, are useful as precursors in the manufacture of other polymeric materials and as additives or components of formulations for plastics, elastomerics, and surface coating compositions, as well as being useful in their own right in many applications. For example, low molecular weight polymers are often required for ease of processing. End-functional polymers are important as building-blocks for advanced copolymers. If of sufficiently low molecular weight, end functional polymers often display useful surface active or compatibilising properties.
- 15 In conventional polymerisation practice, the manufacture of low molecular weight polymers requires the use of an initiator and a chain transfer agent. The initiator acts as a free radical source, whereas the chain transfer agent or regulator controls the molecular weight of the polymer molecule by reacting with the propagating polymer radical to terminate its growth. The chain transfer agent then causes the
- 20 initiation of a new polymer chain thus transferring the growth process from one discrete polymer molecule to another discrete polymer molecule. At least a part of the chain transfer agent is incorporated into the polymer molecule and is thereby consumed in the process.
- 25 The chain transfer agents most commonly used are alkanethiols which possess an objectionable odour, lead to a wide distribution of molecular weights in batch polymerisations with certain monomers, do not allow the production of di-end functional polymers and have limitations as to the types of functional groups that can be installed at the end of the polymer chain. There is also little scope with
- 30 thiols for the chain transfer constant to be optimised for a particular polymerisation. In many polymerisations with thiols, the chain transfer constant, a measure of the effectiveness of the polymerisation regulator, departs significantly

from the ideal of 1.0 that is the optimum for batch polymerisations at moderate to high conversions. The desirability of chain transfer constants close to 1.0 is reviewed by an article by T. Corner in *Advances in Polymer Science*, volume 62, p. 95 (1985). International Patent Application PCT/AU87/00412 and Australian Provisional Patent Application PJ7146/89 disclose novel polymerisation processes employing regulators that help overcome many of the disadvantages of thiols, particularly in relation to chain transfer constant and, in part, provide polymerisation processes that give alternative end groups. They also allow incorporation of a wider variety of useful functional groups at the ends of polymer chains.

The present invention seeks to overcome the disadvantages of polymerisations regulated with thiols by using alternative polymerisation regulators. These regulators have good stability and shelf life while maintaining many of the advantages over thiols. In the majority of cases, the materials that are part of the present process present a different range of chain transfer activities, allowing more opportunity for an optimal process to be selected for a given polymerisation system of monomers and polymerisation conditions. The chain transfer constant that a given regulator possesses is an important consideration in selecting the optimum process for producing low molecular weight polymers.

The present invention provides a process for the production of low molecular weight polymers by free radical polymerisation, which process is characterised by the addition to the polymerisation system of a compound of the general formula I



wherein

R^1 is a hydrogen atom, an alkyl group, or preferably, a group capable of activating the vinylic carbon towards free radical addition: and

R^2 represents an optionally substituted alkyl, optionally substituted alkenyl,

SUBSTITUTE SHEET

optionally substituted alkynyl, or optionally substituted saturated, unsaturated or aromatic carbocyclic or heterocyclic ring;

Suitable groups for R^1 are optionally substituted phenyl or other optionally substituted aromatic groups, alkoxycarbonyl or aryloxy carbonyl ($-\text{COOR}$), carboxy ($-\text{COOH}$), acyloxy ($-\text{O}_2\text{CR}$), carbamoyl ($-\text{CONR}_2$), and cyano ($-\text{CN}$), where R is an
10 alkyl or aryl group.

Optional substituents for R^1 and/or R^2 in formula I may comprise either reactive or non-reactive groups. "Reactive substituent groups" are groups which do not
15 take part in the actual lowering of the molecular weight but are installed at the ends of the polymer chains and may be capable of subsequent chemical reaction. The low molecular weight polymer containing such a reactive group or groups is thereby able to undergo further chemical transformation, such as being joined with
20 another polymer chain. Suitable reactive substituents include hydroxy ($-\text{OH}$), amino ($-\text{NH}_2$), halogen, allyl, cyano, epoxy, and carboxylic acid and its derivatives, such as ester groups ($-\text{COOAlkyl}$). "Non-reactive substituent groups" may be any groups which are not deleterious to the polymerisation reaction or product, for example, alkoxy ($-\text{OAlkyl}$) or alkyl groups.

Substituted rings may have their reactive substituent groups directly attached to
25 the ring or indirectly attached by means of a methylene group or other side chain.

Alkyl groups referred to in this specification may contain from 1 to 32 carbon atoms. Alkenyl and alkynyl groups may contain from 2 to 32 carbon atoms. Saturated, unsaturated, or aromatic carbocyclic or heterocyclic rings may contain
30 from 3 to 14 atoms.

The process of this invention may be adopted by the users of conventional

SUBSTITUTE SHEET

- processes using thiols with little change to reaction conditions other than the substitution of the appropriate quantity of a compound of general formula I for the thiol. The proportion of the compound of general formula I used may be in the range of 0.01 to 30 mole percent based on total monomer, with a preferred range 0.1 to 5 mole percent. The process may be operated at any of the reaction conditions appropriate to free radical polymerisation, i.e. temperatures from -100°C to 400°C and pressures from below atmospheric to 3000 atmospheres. Bulk, solution, emulsion, suspension or other conventional polymerisation modes may be used. Any unsaturated monomers susceptible to free radical polymerisation may be used although it should be noted that the chain transfer constant will vary with the monomer used. Suitable unsaturated monomers include acrylic esters, methacrylic esters, acrylonitrile, vinyl halides, vinyl esters, vinyl aromatics, unsaturated or poly unsaturated hydrocarbons, or mixtures of these. For example, the process is applicable to the manufacture of synthetic rubbers, and other polymer formulations where reduced molecular weight aids processing and improves properties. The process can also be used to produce low molecular weight polymers and oligomers for a variety of applications such as high-solids surface coatings, paints, and adhesives.
- Compounds of general formula I are readily prepared and, unlike the lower molecular weight thiols, they do not possess an objectionable odour. The compounds used in the process of this invention display an unexpected high activity in controlling molecular weight in polymerisation reactions and have chain transfer constants that may be superior to those of thiols, particularly with styrene and acrylates. Their activity is such that their chain transfer constants can approach the optimum of 1.0 for batch polymerisations and this activity is not as highly dependent as that of thiols on the structure of the propagating radical.

- A feature of the process of this invention is that, unlike processes described in many of the examples of International Patent Application PCT/AU87/00412, it produces polymer chains that do not contain terminal unsaturation. In addition, those chain transfer agents described in International Patent Application

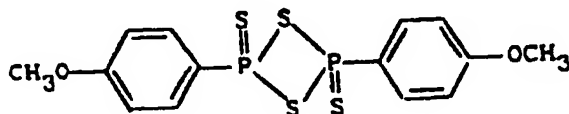
- 5 -

PCT/AU87/00412 that specifically do not introduce terminal unsaturation (for example, α -benzyloxystyrene and derivatives) have less satisfactory chain transfer constants and lower shelf life than the compounds of formula I.

- 5 An additional, unique feature of the process is that the resulting polymers can be hydrolysed to give a terminal thiol group. Such hydrolysis cleaves off the residue of the chain transfer agent containing R^1 and therefore there is no need for R^1 to contain a functional substituent for the production of thiol terminated polymers. If R^2 , however, contains a functional substituent, a di-end functional polymer will
- 10 be the product of the process (after the hydrolysis step) and one of the end groups will be a thiol group. Thiol terminated polymers have application as odour-free polymeric chain transfer agents and as building blocks for the preparation of block and graft copolymers.
- 15 The end functional polymers produced by any of the aspects of the process described above can be converted into polymers with different end functionality by chemical reaction as is well known in the art. For example, hydroxy terminated polymers can be converted into macromonomers by reaction with methacryloyl chloride. These macromonomers are useful materials for the preparation of graft
- 20 copolymers by free radical copolymerisation.

The materials of formula I that are employed in the process of this invention may be prepared by reaction of the corresponding O-ester with Lawesson's reagent or by treatment of alcohols or alkoxides with thiobenzoyl chloride or the sodium salt

25 of (thiobenzylthio)acetic acid. The reaction of iminoesters with hydrogen sulfide has also been reported to give thionoesters.



Lawesson's Reagent

SUBSTITUTE SHEET.

2103595

- 6 -

The following illustrates some of the methods used to prepare examples of the compounds used in the process.

Benzyl thionobenzoate:

- 5 *By method A [Lawesson's reagent].* Benzyl benzoate (2.1 g) was heated at 140 °C for 24 h with Lawesson's Reagent (4.9 g) in anhydrous xylene (10 ml). After removal of the solvent and subjection of the mixture to chromatography on silica gel and recrystallization from pentane at low temperature, benzyl thionobenzoate was obtained in 67% yield. ¹H NMR (CDCl₃): δ = 5.64 (s, 2H), 7.3-8.2 (m, 10H).

10

- By method B [Thionobenzoyl chloride route].* Thionyl chloride (71 g) was added dropwise to a solution of dithiobenzoic acid (47 g) in anhydrous ether (44 ml). After 7 h of stirring at ambient temperature, the ether and excess thionyl chloride were removed under reduced pressure. The residue was then distilled twice to afford the thioacid chloride (61%): b.p. 54-64 °C [0.01 mmHg]. Triethylamine (2.03 g) was added in one portion to a solution of the thioacid chloride (3.13 g) and benzyl alcohol (2.16 g) in dry acetone (40 ml) under nitrogen with vigorous stirring. The stirring was continued for 15 h at ambient temperature. The mixture was then poured into water and extracted with ether. The extracts were washed and dried (MgSO₄) and the residue was subjected to chromatography on silica gel (eluent: hexane) to give a viscous yellow oil (3.0 g) which was crystallised from pentane to afford benzyl thionobenzoate (2.4 g, 53%): m.p. 39-40 °C.

15

20

- By method C [(Thiobenzylthio)acetic acid route].* Phenylmagnesium bromide was prepared by adding bromobenzene (20 g) in ether (100 ml) dropwise under nitrogen and reflux to a stirred mixture of magnesium turnings (3.2 g) in anhydrous ether (50 ml). After 30 minutes boiling under reflux, the mixture was cooled in ice and carbon disulfide (12 ml) was added dropwise. The mixture was allowed to warm gradually to 20 °C and stirred for a further 15 h. After this period, ice (130 g) was cautiously added, and the organic layer was separated. The sodium salt of chloroacetic acid (12 g) was added to the aqueous phase and the mixture was allowed to stand for 24 h. After this period, it was brought to pH

25

30

SUBSTITUTE SHEET

1 with hydrochloric acid and extracted with ether. The ether extracts were washed with water (x 3), dried (CaCl_2), and the solvent was removed to afford a residue that was recrystallised from benzene to give the acid derivative (7.9 g): m.p. 125-126 °C). A portion (1.06 g) was dissolved in dry tetrahydrofuran (125 ml) and
5 sodium hydride (0.48 g) was then added. After the effervescence had ceased, imidazole (0.68 g) was added and the mixture was refluxed for 5 minutes. Benzyl alcohol was then added and the mixture was refluxed for a further 5 min. It was then cooled, poured into water, and extracted with ether. The extracts were washed three times with water, dried and the solvent was removed to give benzyl
10 thionobenzoate (0.72 g), which was further purified by flash chromatography on silica gel (eluent: b.p. 40-60 °C petroleum spirit) and recrystallization from pentane. Yield: 4 g. m.p. 38-39 °C.

(4-Methoxycarbonylbenzyl) thionobenzoate: Method C was used to prepare this
15 compound in low yield (ca. 5% overall) from bromobenzene and methyl (4-hydroxymethyl)benzoate. The thionoester was recrystallised from dichloromethane/hexane: m.p. 90-91 °C. ^1H NMR (CDCl_3): δ = 3.92 (s, 3H), 5.74 (s, 2H), 7.2-7.7 (m, 5H), 7.9-8.3 (m, 4H). MS (CH^+): m/z 287 (MH^+ , 25%), 149 (32%).

20

Benzyl 4-methoxythionobenzoate: This compound was prepared from 4-bromoanisole and benzyl alcohol in 5% overall yield by Method C. It was recrystallised from dichloromethane/hexane: m.p. 68-69 °C. ^1H NMR (CDCl_3): δ = 3.67 (s, 3H), 5.60 (s, 2H), 6.73 (d, J = 9 Hz, 2H), 7.1-7.6 (m, 5H), 8.13 (d, J =
25 9 Hz, 2H). MS (CH^+): m/z 259 (MH^+ , 10%), 135 (100%), 107 (10%), 91 (45%).

(4-Methoxycarbonylbenzyl) 4-methoxythionobenzoate: Method C was used to prepare this compound in 11% overall yield from 4-bromoanisole and methyl (4-hydroxymethyl)benzoate. The crude product was subjected to flash
30 chromatography (eluent: dichloromethane) and recrystallised from dichloromethane/hexane: m.p. 83-85 °C. ^1H NMR (CDCl_3): δ = 3.83 (s, 3H), 3.93 (s, 3H), 5.75 (s, 2H), 6.83 (d, J = 7.5 Hz, 2H), 7.50 (d, J = 7.5 Hz, 2H), 7.9-8.3

2103595

- 8 -

(m, 4H). MS (CH^+): m/z 317 (MH^+ , 3%), 149 (18%), 135 (100%). Accurate mass m/z 317.0836. $\text{C}_{17}\text{H}_{17}\text{O}_4\text{S}$ requires m/z 317.0847.

(4-Ethoxycarbonylbenzyl) 4-methoxythionobenzoate: This compound was prepared from 4-bromoanisole and ethyl (4-hydroxymethyl)benzoate by Method C. The overall yield was 18% and recrystallization from dichloromethane/hexane was used for purification. m.p. 75-77 °C. ^1H NMR (CDCl_3): δ = 1.40 (t, J = 7 Hz, 3H), 3.83 (s, 3H), 4.37 (q, J = 7 Hz, 2H), 5.73 (s, 2H), 6.80 (d, J = 7.5 Hz, 2H), 7.50 (d, J = 7.5 Hz, 2H), 8.0-8.4 (m, 4H). MS (CH^+): m/z 331 (MH^+ , 40%), 163 (40%), 135 (100%).

4-(Hydroxymethyl)benzyl thionobenzoate: This compound was prepared using Method B in 26% yield from thiobenzoyl chloride and 1,4-benzenedimethanol. After recrystallization from hexane, yellow needles of the thionoester were obtained. m.p. 80-80.5 °C. ^1H NMR (CDCl_3): δ = 1.67 (s, 1H), 4.68 (s, 2H), 5.68 (s, 2H), 7.40 (m, 7H), 8.17 (d, J = 6 Hz, 2H). MS (EI): m/z 257 (M^+-1 , 12%), 241 (90%), 121 (100%). Accurate mass m/z 258.0732. $\text{C}_{15}\text{H}_{14}\text{O}_2\text{S}$ requires m/z 258.0714.

The following non-limiting examples illustrate the invention.

EXAMPLE 1

Preparation of Low Molecular Weight Polystyrene with Benzyl Thionobenzoate

Azobisisobutyronitrile (35 mg) was dissolved in freshly distilled styrene (25 ml). Aliquots (5.0 ml) were removed and added to ampoules containing the amount of benzyl thionobenzoate shown below in Table 1. The mixtures were polymerised at 60 °C for 1 h in the absence of oxygen. The contents of the ampoule were then poured into methanol and the precipitated polymer was collected and dried *in vacuo* overnight. A small portion was examined by gel permeation chromatography (GPC) using a Waters Instrument connected to six μ -Styragel columns (10^6 , 10^5 , 10^4 , 10^3 , 500, and 100 Å pore size). Tetrahydrofuran was used

SUBSTITUTE SHEET

as eluent at a flow rate of 1 ml/min and the system was calibrated using narrow distribution polystyrene standards (Waters). The results appear in Table 1.

Table 1

5	Amount of benzyl thionobenzoate added	Conversion %	\bar{M}_n
	81 mg	2.7	11400
10	41 mg	2.9	20800
	21 mg	3.0	37600
	0 mg	3.2	124000

The chain transfer constant (C_x), calculated from these data, was 1.0 which
15 compares favourably with that from n-butanethiol ($C_x = 21-25$). These results show that the compound is an efficient chain transfer agent and that the process produces polymers of low molecular weight in a controlled manner.

EXAMPLE 2

20 Preparation of Low Molecular Weight Poly(methyl acrylate) with Benzyl Thionobenzoate

Azobisisobutyronitrile (9 mg) was dissolved in freshly distilled methyl acrylate (25 ml). Aliquots (2.0 ml) were removed and added to ampoules containing thiophen-
25 free benzene (8 ml) and the amount of benzyl thionobenzoate shown below in Table 2. The mixtures were polymerised at 60°C for 1 h in the absence of oxygen. The volatiles were then removed and the polymers were dried *in vacuo* to constant weight and then examined as before. The results appear in Table 2.

2103595

Table 2

	Amount of benzyl	Conversion	\bar{M}_n *
	thionobenzoate added	%	
5			
	50 mg	3.8	9020
	26 mg	6.0	17700
	12 mg	9.2	35600
	0 mg	16	473000
10			

* Polystyrene-equivalent number average molecular weight, obtained by GPC.

The chain transfer constant (C_x), calculated from these data, was 1.2, which
15 compares favourably with that from n-butanethiol ($C_x = 1.7$).

EXAMPLE 3

Preparation of Low Molecular Weight Poly(vinyl acetate) with Benzyl Thionobenzoate

20 Azobisisobutyronitrile (16 mg) was dissolved in freshly distilled vinyl acetate (100 ml). Aliquots (10.0 ml) were removed and added to ampoules containing the amount of benzyl thionobenzoate shown below in Table 3. The mixtures were polymerised at 60 °C for 1 h in the absence of oxygen. The volatiles were then removed and the polymers were dried *in vacuo* to constant weight and then
25 examined as before. The results appear in Table 3.

Table 3

5	Amount of benzyl	\bar{M}_n *
	thionobenzoate added	
	2.2 mg	1900
	0	253000

* Polystyrene-equivalent number average molecular weight, obtained by GPC.

10

The chain transfer constant (C_{tr}), calculated from these data, was > 20 . These results show that the compound acts as a very active regulator in polymerisation of vinyl acetate.

15 **EXAMPLE 4**

Preparation of Low Molecular Weight Polystyrene with (4-Methoxycarbonylbenzyl) Thionobenzoate

20 Samples of polystyrene were prepared on the same scale and in the manner of example 1. The amount of (4-methoxycarbonylbenzyl) thionobenzoate added and the results of the polymerisation are shown in Table 4.

2103595

- 12 -

Table 4

5	Am unt of (4-methoxycarbonylbenzyl) thionobenzoate added	\bar{M}_n
	80 mg	23000
	40 mg	41900
	20 mg	66300
	0 mg	142000
10		

The chain transfer constant (C_{tr}), calculated from these data, was 0.59. A sample of low molecular weight polystyrene ($\bar{M}_n = 4570$) prepared with (4-methoxycarbonylbenzyl) thionobenzoate was examined by ^1H NMR

- 15 spectroscopy and showed signals at $\delta = 3.83$ indicative of the presence of methyl ester groups. Integration of the spectrum and comparison with the integral of the aromatic styrene signal showed there to be an end group functionality of 0.9-1.0. This experiment shows that the process can be used to prepare end functional polymers.

20

EXAMPLE 5

Preparation of Low Molecular Weight Poly(methyl acrylate) with (4-Methoxycarbonylbenzyl) Thionobenzoate

- 25 Samples of poly(methyl acrylate) were prepared according to the directions given in Example 2, except that (4-methoxycarbonylbenzyl) thionobenzoate was used rather than benzyl thionobenzoate. The amount of (4-methoxycarbonylbenzyl) thionobenzoate added and the results of the polymerisation are shown in Table 5.

Table 5

5	Amount of (4-methoxycarbonylbenzyl) thionobenzoate added	\bar{M}_n *
	50 mg	7700
	26 mg	14300
	12 mg	28300
	0 mg	772000
10		

* Polystyrene-equivalent number average molecular weight, obtained by GPC

The chain transfer constant (C_x), calculated from these data, was 1.4, which compares favourably with that from n-butanethiol ($C_x = 1.7$).

15

EXAMPLE 6

*Preparation of Low Molecular Weight Polystyrene with Benzyl
4-Methoxy(thionobenzoate)*

20 Samples of polystyrene were prepared on the same scale and in the manner of example 1. The amount of benzyl 4-methoxy(thionobenzoate) added and the results of the polymerisation are shown in Table 6.

2103595

- 14 -

Table 6

	Amount of benzyl	\bar{M}_n
	4-methoxy(thionobenzoate) added	
5		
	80 mg	65100
	40 mg	93500
	20 mg	113000
	0 mg	145000
10		

The chain transfer constant (C_x), calculated from these data, was 0.12.

EXAMPLE 7

Preparation of Low Molecular Weight Poly(methyl acrylate) with Benzyl

15 *4-Methoxy(thionobenzoate)*

Samples of poly(methyl acrylate) were prepared according to the directions given in Example 2, except that benzyl 4-methoxy(thionobenzoate) was used rather than benzyl thionobenzoate. The amount of benzyl 4-methoxy(thionobenzoate) added and the results of the polymerisation are shown in Table 7.

Table 7

	Amount of benzyl	\bar{M}_n *
	4-methoxy(thionobenzoate) added	
25		
	50 mg	10700
	26 mg	18600
	13 mg	42600
30	0 mg	394000

* Polystyrene-equivalent number average molecular weight, obtained by GPC

SUBSTITUTE SHEET

The chain transfer constant (C_x), calculated from these data, was 1.1, which compares favourably with that from n-butanethiol ($C_x = 1.7$).

EXAMPLE 8

5 Preparation of Low Molecular Weight Polystyrene with 4-(Methoxycarbonyl)benzyl 4-Methoxy(thionobenzoate)

Samples of polystyrene were prepared on the same scale and in the manner of example 1. The amount of 4-(methoxycarbonyl)benzyl 4-methoxy(thionobenzoate) added and the results of the polymerisation are shown in Table 8.

Table 8

	Amount of 4-(methoxycarbonyl)benzyl	\bar{M}_n
15	4-methoxy(thionobenzoate) added	
	81 mg	48400
	40 mg	86000
	20 mg	105300
20	0 mg	150000

The chain transfer constant (C_x), calculated from these data, was 0.25. The total average functionality (methoxy and methoxycarbonyl groups) calculated by ^1H NMR on a sample of polystyrene of $\bar{M}_n = 13300$ (prepared with 4-(methoxycarbonyl)benzyl 4-methoxy(thionobenzoate) was 2.1, which shows that 4-(methoxycarbonyl)benzyl 4-methoxy(thionobenzoate) is efficient in introducing functional groups at the termini of polymer chains and that this type of chain transfer agent can be used to prepare end-functional polymers.

30 EXAMPLE 9

Preparation of Low Molecular Weight Poly(methyl acrylate) with 4-(Methoxycarbonyl)benzyl 4-Methoxy(thionobenzoate)

Samples of poly(methyl acrylate) were prepared according to the directions given in Example 2, except that 4-(methoxycarbonyl)benzyl 4-methoxy(thionobenzoate) was used rather than benzyl thionobenzoate. The amount of 4-(methoxycarbonyl)benzyl 4-methoxy(thionobenzoate) added and the results of the polymerisation are shown in Table 9.

Table 9

10	Amount of 4-(methoxycarbonyl)benzyl 4-methoxy(thionobenzoate) added	\bar{M}_n *
	50 mg	10700
	26 mg	18600
	13 mg	42600
15	0 mg	394000

* Polystyrene-equivalent number average molecular weight, obtained by GPC

The chain transfer constant (C_x), calculated from these data, was 1.1, which compares favourably with that from n-butanethiol ($C_x = 1.7$).

EXAMPLE 10

Preparation of Low Molecular Weight Polystyrene with 4-(Ethoxycarbonyl)benzyl 4-Methoxy(thionobenzoate)

25

Samples of polystyrene were prepared on the same scale and in the manner of Example 1. The amount of 4-(ethoxycarbonyl)benzyl 4-methoxy(thionobenzoate) added and the results of the polymerisation are shown in Table 10.

Table 10

	Amount of 4-(ethoxycarbonyl)benzyl 4-methoxy(thionobenzoate) added	\bar{M}_n
5	616 mg	13500
	0 mg	150000

10 The presence of ethoxycarbonyl and methoxy end groups in the lower molecular weight sample of polystyrene was shown by signals in the ^1H NMR spectrum at $\delta = 4.28$ and 3.83 , respectively.

EXAMPLE 11

15 *Preparation of Low Molecular Weight Polystyrene with 4-(hydroxymethyl)benzyl thionobenzoate*

20 Samples of polystyrene were prepared on the same scale and in the manner of example 1. The amount of 4-(hydroxymethyl)benzyl thionobenzoate added and the results of the polymerisation are shown in Table 11.

Table 11

	Amount of 4(hydroxymethyl)benzyl thionobenzoate added	\bar{M}_n
25	81 mg	25600
	40 mg	43900
	20 mg	59500
	0 mg	113000

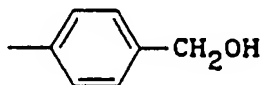
30 The chain transfer constant was 0.43. The presence of the end group shown below in a polymer of $\bar{M}_n = 6810$, prepared with 4-(hydroxymethyl)benzyl

2103595

- 18 -

thionobenzoate, was confirmed by a broad signal in the ^1H NMR spectrum at $\delta = 4.4\text{--}4.7$ (due to the benzyl methylene hydrogens) and an infrared absorption at 3415 cm^{-1} .

5



End group of polymer prepared with 4-(hydroxymethyl)benzyl thionobenzoate

- 10 After treatment with *t*-butyldimethylsilyl chloride and imidazole, this polymer showed signals at $\delta = 0.08$ in the ^1H NMR spectrum due to the methyls of a *t*-butyldimethylsilyl ether group. The formation of this silyl ether further confirms the presence of a hydroxyl end group.

SUBSTITUTE SHEET



Canadian Intellectual
Property Office

Off. de la propriété
intellectuelle du Canada

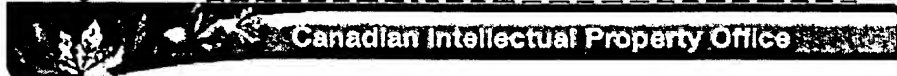
Canada

Français	Contact Us	Help	Search	Canada Site
Strategis	Site Map	What's New	About Us	Registration

strategis.gc.ca



Strategis Index: [A](#)[B](#)[C](#)[D](#)[E](#)[F](#)[G](#)[H](#)[I](#)[J](#)[K](#)[L](#)[M](#)[N](#)[O](#)[P](#)[Q](#)[R](#)[S](#)[T](#)[U](#)[V](#)[W](#)[X](#)[Y](#)[Z](#)



Canadian Patents Database

(12) Patent Application:

(11) CA 2127919

(54) PROCESS FOR PRODUCING ULTRAFINE SIZED LATEXES

(54) PROCEDE DE PRODUCTION DE LATEX ULTRAFINS

[View or Download Images](#)

[View Administrative Status](#)

ABSTRACT:

ABSTRACT OF THE DISCLOSURE

A process for producing an aqueous-based dispersion containing between about 15 and about 50 percent by weight solids comprising the steps of:

- (a) incrementally adding one or more ethylenically unsaturated monomers capable of polymerizing in an aqueous environment to a reaction vessel containing water and up to 6.3 parts per hundred parts monomers of one or more surfactants;
- (b) incrementally adding one or more polymerization initiators to said reaction vessel; and
- (c) allowing said one or more ethylenically unsaturated monomers to polymerize such that the average particle size of said polymerized monomers is less than 100 nanometers; and the dispersion produced therein is disclosed. The produced dispersions have a wide range of practical uses.

CLAIMS: [Show all claims](#)

*** Note: Data on abstracts and claims is shown in the official language in which it was submitted.